The sweet spot of forward osmosis: Treatment of produced water, drilling wastewater, and other complex and difficult liquid streams

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HIGHLIGHTS

• Highly impaired liquid streams can be sustainably treated by forward osmosis.
• Forward osmosis treatment of drilling mud and produced water was evaluated.
• Water recovery > 70% was achieved in pilot and demonstration scales.

GRAPHICAL ABSTRACT

ABSTRACT

Global water scarcity and substantial challenges associated with treatment of complex and impaired liquid streams have advanced the development of forward osmosis (FO), which can successfully treat and recover water for beneficial reuse. Surging research and advancements in FO, a technology once unable to compete with conventional wastewater treatment processes, have identified its sweet spot: treatment and desalination of complex industrial streams, and especially oil and gas (O&G) exploration and production wastewaters. High salt concentrations, decentralized and transient operations, the presence of free and emulsified hydrocarbons, silts and clays leached from producing formations, and process additives common in O&G drilling wastewater and produced water render many common treatment technologies ineffective. Treatment and reuse of O&G wastewater, and other complex industrial streams, in a cost effective and environmentally sound manner is critical for sustainable industrial development and to meet increasingly stringent regulations. This review focuses on the successful development and demonstration of FO membrane treatment systems, supported by a review of bench-scale, pilot, and demonstration studies on treatment of O&G waste streams, landfill leachates, centrate from anaerobic digesters, activated sludge in membrane bioreactors, and liquid foods and beverages. Recent developments in membrane fabrication, system configurations, and draw solutions are briefly reviewed.

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1. Introduction

The United States and many countries around the world are rapidly expanding exploration and development of unconventional gas resources, including shale gas, coaled methane, and tight sands [1–5]. With recent advancements in horizontal well drilling and hydraulic fracturing, unconventional gas is expected to account for nearly 45% of the natural gas produced in the U.S. by 2035 [6,7]. As production increases and new formations become economically viable, water demands for well development and the volume of wastewater generated during exploration and production (E&P) (e.g., drilling muds, hydraulic fracturing flowback water, produced waters) will increase significantly.

Drilling mud is an integral part of well development, providing lubrication to drilling equipment, stabilization to well walls, pressure control within the borehole, and flushing of debris from the well. Up to one million gallons (3800 m³ or 24,000 bbl) of fresh water can be consumed during drilling of a single well, producing grit-laden streams contaminated with drilling additives and containing high concentrations of chemical oxygen demand (COD), total dissolved solids (TDS), and organic and inorganic constituents [8–11]. When borehole drilling is completed, the drilling mud is usually stored on-site in lined ponds/pits. In some locations, closed-loop drilling is required in which no pits are used. In most drilling operations, these fluids receive minimal treatment and are trucked off-site for deep well injection. Occasionally, the waste fluids may be land applied if a proper permit is obtained [5].

After drilling, well productivity can be enhanced with hydraulic fracturing. Between one and four million gallons (3800–15,000 m³ or 24,000–95,000 bbl) of water-based slurries are injected into the wellbore under high pressure, forming fractures in the target formation [9,12,13]. Hydraulic fracturing facilitates free flow of oil and gas; thus, increasing recovery from formations previously considered economically unfavorable. A portion of the fracturing fluids that were injected is recovered from the well over a span of several weeks, generating a waste stream of water, sand, and chemical additives [7,13]. Depending on the formation, the flowback wastewater can also have high concentrations of TDS attributed to leaching of earth minerals from the subsurface. Similar to drilling muds, fracturing flowback is recovered and stored on-site. Historically, most flowback water received minimal treatment before being disposed into deep wells [7,9,13]; however, Class II injection wells are not available in all locations. Wastewater treatment is possible, and the treated water can supplement or replace the fresh water necessary for drilling and fracturing of additional wells; yet, highly saline wastewater streams and some hydraulic fracturing chemical additives are difficult to treat with conventional processes.

The wastewater stream flowing with the gas after most of the fracturing water is recovered, is considered produced water [13,14]. This stream can represent nearly 70% of the total wastewater generated during the lifetime of a well, producing volumes several times greater than the volume of oil and/or gas recovered [15]. The quantity of produced water is highly dependent on well location, and its quality just as variable. These streams typically contain a wide range of TDS concentration, free and emulsified hydrocarbons, and silt and clay leached from the formation [8,16]. Depending on the quality and composition of produced water, a broad range of technologies can be utilized for its treatment; however, the complexity and total cost of treatment is dependent on its salinity and ultimate use [9].

As the development of unconventional oil and gas (O&G) continues in the U.S. and abroad, maximizing water resources while minimizing the volumes of E&P waste will become increasingly important. Several O&G exploration regions are considered at high risk for water resource depletion [8], providing an excellent opportunity for beneficial reuse of reclaimed waste streams. Properly applied management techniques and emerging water treatment processes can drastically reduce industrial water demands, promoting closed loop water recycling and minimizing environmental exposure associated with E&P of unconventional O&G resources.

Many other industrial streams are difficult to manage, similar to O&G E&P wastewaters, and require special technologies to provide sufficient treatment. For example, landfill leachates are heavily contaminated waste streams that often require advanced treatment processes to provide adequate contaminant rejection prior to discharge or reuse. Water recovery from domestic wastewater sludge and anaerobic digestor centrate has also gained attention as a result of surging interest in direct and indirect potable water reuse in the United States. The nexus between food production and water recovery has also grown in complexity as the food industry strives to increase liquid food and beverage quality, while simultaneously concentrating these streams. Though each stream is unique and complex, O&G wastewater and other industrial streams can be treated by a small group of advanced processes.

2. Processes for treatment of O&G E&P wastewaters

Chemical, biological, and physical processes have been previously investigated and implemented for treatment of O&G E&P wastewaters; however, high salinity, prohibitive capital cost, extreme chemical demand, large installation footprint, residual (brines and solids) management challenges, and limited removal of emerging contaminants are some of the hurdles to successful implementation of many technologies. Desalination processes such as distillation and membrane separation processes, have demonstrated the ability to achieve adequate treatment of these streams; yet, further improvements to these technologies to reduce the high costs and operational challenges, and development of more effective pretreatment are needed before they are broadly adopted and implemented [10,11,15–18].

2.1. Commercial desalination processes

2.1.1. Distillation

In distillation a feed stream is heated and sometimes also placed under partial vacuum to increase its vapor pressure and form water vapor that can be condensed and recovered as high quality water. Vapor extraction can be repeated several times in the process to enhance evaporation while further concentrating the feed stream. Common commercial distillation methods include multi-effect distillation (MEF), multi-stage flash (MSF), and vapor compression (VC) distillation [19]. Desalination by distillation can minimize physical and chemical treatment and the amount of de-oiling equipment necessary for treatment of O&G wastewater. This eliminates capital costs and minimizes secondary chemical waste sludge [17]. Additionally, distillation can treat highly saline feed streams because it is not affected by the high osmotic pressure of saline and hypersaline streams; however, corrosion and scaling can occur during distillation and incur high operating and maintenance (O&M) costs [14,19]. If volatile organic compounds are present in the feed stream, they may be poorly removed because they will volatilize and condense in the distillate stream. Energy demand is also a limiting factor in distillation, accounting for more than 95% of the total operating costs in a recent review of commercial scale processes [17].

2.1.2. Membrane separation

Membrane separation technologies are commonly pressure driven separation processes that rely on diffusive- or convective-based mass transfer phenomena to separate dissolved and suspended constituents from aqueous solutions. Traditional pressure driven membrane technologies include microfiltration (MF), ultrafiltration (UF), nanofiltration (NF), and reverse osmosis (RO). Membrane permeability and the size of constituents rejected by each process decrease in the order presented (MF > UF > NF > RO); while MF sieves suspended particles, RO can effectively reject monovalent ions, including sodium, chloride, and low molecular weight organic compounds [17]. Membrane processes, and especially NF and RO, can successfully reject a broad range of contaminants and TDS present in impaired feed streams.
RO and NF are very effective desalination processes; however, they are highly susceptible to inorganic scaling and to particulate, biological, and organic fouling [20]. These foulants can become compacted and difficult to clean, leading to low water permeability, increased pressure loss, and considerable chemical consumption for cleaning. Additionally, polymeric membranes can be sensitive to feed stream chemical and oil contaminants and natural polymers such as guar (used in the hydraulic fracturing process), which can compromise membrane performance and surface chemistry. Hydraulically driven membrane processes must also overcome the osmotic pressure of the feed stream, limiting the variety of streams (e.g., TDS concentration lower than 70,000 mg/L) that can be treated.

2.2. Engineered osmosis: forward osmosis

Engineered osmosis, and specifically forward osmosis (FO), is an emerging desalination and treatment technology that can provide robust and modular treatment, reject contaminants found in O&G waste streams, and avoid the drawbacks of pressure driven membrane processes. Engineered osmosis is a promising alternative for difficult to treat waste streams such as produced water [14], hydraulic fracturing flowback water, and drilling mud. In some cases, FO can be used as a standalone desalination process, or it can be considered an advanced pretreatment process for RO or NF. The following sections provide details on the principles of FO and showcase its successful treatment of complex industrial wastewater streams.

2.2.1. Principal of forward osmosis

Osmosis is the net transfer of water across a semi-permeable membrane resulting from an osmotic pressure difference across a semipermeable membrane. In FO (Fig. 1a), a synthetic polymeric membrane separates a feed stream and a concentrated draw solution, and the osmotic pressure difference across the membrane facilitates diffusion of water through the membrane from the low osmotic pressure feed to the high osmotic pressure draw solution while rejecting almost all dissolved and suspended constituents [23,24]. Commonly, the FO process is completed in two separate steps: 1) recovery of water from a feed stream and dilution of the draw solution, and 2) production of high quality product water using RO or distillation while reconcentrating the draw solution (Fig. 1b) [18,25–32]. The reconcentrated draw solution is then reused in the FO process; however, several industrial applications such as O&G well fracturing are able to beneficially use the dilute draw solution, eliminating the need for the reconcentration step.

FO has many advantages over other membrane technologies. High rejection of almost all solutes and suspended solids while operating at very low hydraulic pressures and ambient temperature are the greatest benefits of FO. These significantly reduce energy consumption and capital costs associated with pumping and system design and construction. They also allow for the development of highly modular systems that can be operated in harsh conditions with minimal access to electric power and supplies. FO experiences less membrane fouling compared to pressure driven membrane processes such as UF, NF, and RO [33–35]. This is due to minimal cake layer formation and lower compaction of foulants on the membrane active layer. Fouling deposits on FO membranes are easily removed with osmotic backwashing [36–40] or turbulent flow at the feed-membrane interface. During osmotic backwashing, the draw solution is replaced with deionized or fresh water. This develops an osmotic pressure gradient in the opposite direction across the FO membrane and water permeates from the draw solution channel into the feed channels. The permeation of water back into the feed channels helps to dissolve and detach foulants from the membrane surface. Unlike pressure driven membrane processes, FO can be used to treat highly saline feed streams because it does not require high hydraulic pressures to overcome high osmotic pressures.

2.2.2. Draw solutions

Draw solution selection is important for maintaining a sustainable and efficient FO process. There are several factors that dictate what constitutes an appropriate draw solution, which are defined by the type of FO application. If the draw solution requires reconcentration, the chosen solutes should be highly soluble to avoid scaling during RO

Fig. 1. A synthetic polymeric membrane separates a feed stream and a concentrated draw solution, and the osmotic pressure difference across the membrane facilitates diffusion of water through the membrane: (a) the osmosis process and (b) FO process when the osmosis process is coupled with a brine reconcentration system, producing high quality product water, while reconcentrating the FO draw solution to maintain constant osmotic pressure.
or distillation recondition. Draw solutions must also be suitable for industrial applications and inexpensive. Several laboratory studies have investigated suitable draw solutions for their osmotic pressure, recoverability, and mass transfer through the membrane [23,25,28,41–47]. These include mono and divalent salts, dissolved gasses, sugars, engineered nanoparticles, or fertilizers. A review of promising draw solutions was recently published [48], and the osmotic pressures of potential inorganic draw solutions as a function of their molar concentration are presented in Fig. 2 [23].

### 2.2.3. FO membranes

FO membranes have unique properties that enable efficient diffusion of water through the polymer. These include very thin active and support layers and very porous support layer with pores having low tortuosity. Despite being a relatively new process, several manufacturers are developing and commercializing suitable FO membranes (Table 1). The most common commercially available membranes are cellulose triacetate (CTA) and polyamide thin-film composite (TFC) membranes manufactured by Hydration Technology Innovations (HTI, Albany, OR) and Oasys Water (Boston, MA) [48]. Between these companies, different FO membrane packaging configurations have been developed, including plate- and frame, spiral wound, and tubular (e.g., hollow fibers) [23]. Spiral wound FO elements are similar to commercial RO membrane elements; however, they are modified to allow forced-flow inside the membrane envelopes (Fig. 10) [23]. Plate and frame configurations use flat sheet membranes separated by spacers, providing lower surface area to volume ratio in cassette packages. Tubular and hollow fiber FO membranes are commonly potted in large bundles, significantly increasing the packing density of a membrane element.

### 3. Forward osmosis for treatment of complex streams

Through extensive research and development in recent years, FO has been demonstrated as a promising technology for treatment of challenging liquid streams. Successful applications include desalination of seawater and brackish water, concentration of landfill leachates, treatment of wastewater (including in osmotic membrane bioreactor configurations), and processing of foods and beverages [23,25,41]. FO has been investigated at almost all scales as a hybrid pretreatment process for production of high quality water, and as a standalone process where the diluted draw solution is beneficially used. In the following section we summarize various applications where FO was successfully tested for treatment of complex streams, starting with the treatment of waste streams in the O&G industry.

![Figure 2: Osmotic pressure as a function of draw solution concentration for potential draw solutions. Adapted from [23].](image)

#### Table 1

Current FO membrane manufacturers and commercial status. Adapted from [48].

<table>
<thead>
<tr>
<th>Firm/manufacturer</th>
<th>Membrane type</th>
<th>System supply</th>
<th>Primary current application</th>
<th>Commercial status</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aquaporin A/S</td>
<td>Aquaporin</td>
<td>No</td>
<td>FO, OC</td>
<td>Pre-commercial</td>
</tr>
<tr>
<td>Fuji</td>
<td>NA</td>
<td>No</td>
<td>NA&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Development</td>
</tr>
<tr>
<td>GKSS</td>
<td>Polymeric</td>
<td>No</td>
<td>NA&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Development</td>
</tr>
<tr>
<td>GreenCentre Canada</td>
<td>NA&lt;sup&gt;b&lt;/sup&gt;</td>
<td>No</td>
<td>SWFO&lt;sup&gt;c&lt;/sup&gt;</td>
<td>Development</td>
</tr>
<tr>
<td>HTI</td>
<td>CA, TFC</td>
<td>Yes</td>
<td>Various</td>
<td>Commercial</td>
</tr>
<tr>
<td>Idaho National Lab</td>
<td>NA&lt;sup&gt;b&lt;/sup&gt;</td>
<td>No</td>
<td>NA&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Development</td>
</tr>
<tr>
<td>IDE Technologies</td>
<td>NA&lt;sup&gt;b&lt;/sup&gt;</td>
<td>Yes</td>
<td>PRO&lt;sup&gt;d&lt;/sup&gt;</td>
<td>Pre-commercial</td>
</tr>
<tr>
<td>Modern Water</td>
<td>Undefined</td>
<td>Yes</td>
<td>SWFO&lt;sup&gt;c&lt;/sup&gt;</td>
<td>Commercial</td>
</tr>
<tr>
<td>Oasys Water</td>
<td>TFC</td>
<td>Yes</td>
<td>Brine concentration</td>
<td>Commercial</td>
</tr>
<tr>
<td>Porifera</td>
<td>TFC</td>
<td>Yes</td>
<td>Various</td>
<td>Pre-commercial</td>
</tr>
<tr>
<td>Samsung</td>
<td>NA&lt;sup&gt;b&lt;/sup&gt;</td>
<td>No</td>
<td>NA&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Development</td>
</tr>
<tr>
<td>Trevi Systems</td>
<td>NA&lt;sup&gt;b&lt;/sup&gt;</td>
<td>Yes</td>
<td>SWFO&lt;sup&gt;c&lt;/sup&gt;</td>
<td>Development</td>
</tr>
</tbody>
</table>

<sup>a</sup> Demonstration-scale FO membrane treatment systems available (yes/no).
<sup>b</sup> Not available.
<sup>c</sup> Seawater forward osmosis.
<sup>d</sup> Pressure retarded osmosis.
<sup>e</sup> Osmotic concentration.

#### 3.1. FO treatment of O&G E&P wastewater

#### 3.1.1. The green machine

The three dominant waste streams generated during E&P contain chemicals and polymers that assist in drilling and hydraulic fracturing, minerals leached from the formations, and organic and inorganic constituents. Research conducted by Hutchings et al. [11] investigated the performance of FO for treatment of O&G waste streams for beneficial, intra-basin reuse using the Green Machine concept.

The Green Machine is a mobile and scalable FO treatment process operated at the well site, thus limiting water and wastewater trucking and providing a local, reusable water source. Operational success was measured by the system’s ability to minimize fresh water demand through reuse, preventing secondary waste generation, reducing O&G wastewater volumes for disposal, and utilizing readily available and on-site chemical energy to generate a predetermined osmotic pressure driving force. Carbon footprint reduction, resulting from efficient FO operation and minimized trucking demands, was also of high importance.

Since 2010, two distinct models of the Green Machine have been manufactured and pilot tested by HTI and Emerald Surf Sciences (Shreveport, LA) (previously Bear Creek Services). The first generation Green Machine (Fig. 3) utilized 20 to 280 vertically oriented, 8-inch (0.2 m) diameter by 40-inch (1.0 m) long spiral wound FO membrane elements to treat stream flows of 8 to 170 gal/min (30–640 L/min; 275–5800 bbl/d). The system operated in an osmotic dilution mode, where a 26% w/w NaCl draw solution recirculates inside the membrane envelope while drilling wastewater (2.5% w/w TDS) flows by gravity on the active side of the membrane. The highly concentrated NaCl draw solution is diluted to less than 7% w/w (~70,000 mg/L) NaCl while concentrating the E&P wastewater by more than 3.5 times (greater than 70% water recovery). Testing results [49] showed that this system was able to reclaim more than 125,000 gallons (473 m<sup>3</sup> or 3000 bbl) of O&G wastewater using less than twenty gallons (75 L) of diesel fuel. This same volume would have required over 25 truckloads for disposal at an off-site deep well injection facility. The first generation system could ultimately save nearly a million gallons (3800 m<sup>3</sup> or 24,000 bbl) of water per well application and account for up to 20% of the saline completion fluid needed at each drilling location. These savings translate into approximately 150 saved truckloads, both in fresh water and fuel consumption.

However, in a recent study [10], results have shown that the first generation Green Machine FO treatment could be further optimized.
Using a custom made FO membrane test cell with a CTA membrane and a 26% w/w (~320,000 mg/L) NaCl draw solution, osmotic dilution experiments were performed during which at least 50% of the O&G drilling waste feed volume was recovered. Results from the study suggest that while FO can concentrate O&G drilling waste streams by up to three times, increased feed stream velocities can decrease membrane fouling and concentration polarization [50], minimize feed channel clogging, and leads to higher water flux. Minimal irreversible fouling was also observed, demonstrating the effectiveness of osmotic backwashing of FO membranes [36–40]. Additionally, high rejection of inorganic and organic constituents during pilot testing was confirmed by the bench-scale study.

In 2012 the second generation Green Machine (Fig. 4) was developed, optimizing system performance as a result of previous experimental and pilot testing. The second generation FO system utilizes 24 horizontally oriented, 8-inch diameter by 40-inch long spiral wound FO membrane elements housed in pressure vessels on a mobile membrane skid. The system operates under forced feed flow through the membrane elements (~40 to 60 PSI hydraulic transmembrane pressure) and is coupled with an RO system for reconcentration of the NaCl draw solution.

Recent pilot testing was conducted where 6% w/w (~60,000 mg/L) NaCl draw solution was diluted to 4.5% w/w (~45,000 mg/L) after a once through pass in the FO system. The diluted draw solution was then reconcentrated by the RO system, producing a high quality permeate stream. During a weeklong field test in the Haynesville shale gas play, the system recovered 85% of O&G drilling wastewater permeate stream. During a weeklong period using the first generation Green Machine, where the diluted draw solution was suitable for use as completion fluid at future drilling applications, the second generation Green Machine produces a high quality RO permeate suitable for a wide range of reuse applications. However, this comes at the price of increased operating and pumping costs. The second generation Green Machine is also limited to a maximum 7% w/w NaCl draw solution if RO reconcentration is used.

3.1.2. The FO membrane brine concentrator (MBC)

Oasys Water has developed the first Membrane Brine Concentrator (MBC) (Fig. 5) employing a patented ammonia/carbon dioxide based draw solution to treat high salinity brine streams and O&G wastewater. A fully integrated, mobile, demonstration scale Oasys MBC system was constructed and field-tested treating high salinity, pretreated produced water. The mobile treatment system consists of three components: pretreatment, MBC platform, and product water/brine polishing. Raw produced water is first pretreated in a chemical reactor where chemical oxidizer, caustic soda, and soda ash are added to form mineral precipitates and organic flocs. The precipitate suspension is pumped through a filter press to separate the sludge from the treated raw water. The pretreated produced water is then filtered through a greensand media filter for additional iron and particulate removal and then through a cartridge filter. Pretreated feed water flows into the FO membrane and is concentrated by the MBC to between 150,000 and 250,000 mg/L TDS, depending on subsystem operating conditions [18].

The proprietary draw solution is a mixture of ammonium bicarbonate and ammonium hydroxide dissolved in water. The resulting draw solution is highly soluble and produces a high osmotic pressure driving force that facilitates permeation of water through the TFC membrane even when the salinity of the feed stream exceeds 200,000 mg/L TDS.
The diluted draw solution is then heated to evaporate the thermolytic draw solution solutes, which have lower vapor pressure than water. This recovery method requires less energy than would be required to overcome the enthalpy of vaporization of water during conventional distillation [51]. The ammonia and carbon dioxide gasses are then condensed, and a reconcentrated draw solution is generated for reuse in the FO system. During piloting of the MBC process the system retained more than 99.75% of its nitrogen containing species during 100 h of operation [18]. Product water stripped of dissolved ammonia and carbon dioxide exits the system as a purified water stream.

In two separate commercial demonstrations (Fig. 6), the MBC process demonstrated to provide water treatment and waste volume minimization of fracturing flowback and produced water from the Marcellus Shale and Permian Basin. During the Marcellus Shale trial, approximately 60,000 gallons (230 m³ or 1430 bbl) of produced water were treated during 800 h of operation. Pilot operations were sustained for a six-month period, and included seven weeks of continuous (5 days a week/24 h a day) operation. Average daily steady-state water flux was between 2 and 3 L/m² h (LMH) depending on operating conditions of the system (i.e., draw solution and feed temperature, draw solution concentration, and solution flow rates). It is important to note that water flux under these conditions (feed TDS concentration between effectively 6.5% and 7.5% w/w NaCl) with hydraulically driven membrane based processes would be negligible, if not negative, due to operating limits and material constraints of these systems [52]. System water recovery averaged 64%.

During system operations in the Permian Basin, approximately 40,000 gallons (150 m³ or 950 bbl) of produced water were treated during 400 h of operation. Average MBC feed salinity in the Permian Basin was 103,000 ± 7,000 mg/L and contained a high concentration of TOC, boron, and heavy metals. Although TDS and hardness were relatively constant, organic and heavy metal constituents and their concentration were observed to vary substantially between wastewater batches. The average TDS concentration of treated water from the MBC in the Permian Basin trial was 737 ± 284 mg/L and the concentrated brine concentration averaged 241,000 ± 35,000 mg/L TDS. System water flux averaged 3 L/m² h, and average system recovery was 60% [18]. The higher water flux and percent recovery despite the higher salinity of the feed stream compared to the Marcellus Shale demonstration are attributable to improvements in the Oasys membrane, membrane element, and other subsystem refinements.

In both the Marcellus Shale and the Permian Basin demonstrations, it is evident that the MBC system is capable of achieving substantial water recovery from highly saline brines, thereby minimizing brine disposal volumes and generating a high quality, tunable product water quality suitable for numerous beneficial use applications.

3.2. Other applications of FO for difficult waste streams

3.2.1. FO treatment of landfill leachate

Most landfills produce leachate, which originates from decomposition of stored wastes or from precipitation that percolates through the piled solid waste. Typical contaminants of concern in landfill leachates include TDS, dissolved metals, organic matter, and organic/inorganic
nitrogen. The volume and concentrations of leachate constituents can be highly variable and they depend on the location of the landfill and corresponding local climate. Leachates are commonly sent to conventional wastewater treatment facilities; however, TDS present in leachates is not efficiently removed by conventional wastewater treatment processes and it can negatively impact biological processes and effluent quality [23].

An FO pilot study was conducted at the Coffin Butte Landfill in Corvallis, Oregon in 1998, attempting to provide advanced treatment of leachate [53]. The landfills are located in the Pacific Northwest and receives enough precipitation to produce 20,000–40,000 m³ of leachate annually (annual average of 15,000–30,000 gal/day). In this particular case, the leachate had to be treated for surface discharge with effluent TDS concentration of less than 100 mg/L.

Cellulose triacetate membranes from Osmotek (now HTI) were utilized for the three-month pilot study. Using NaCl draw solution, the pilot system was operated at 94–96% water recovery, while providing high contaminant rejection and minimal irreversible membrane fouling [23]. As a result of successful piloting, a full-scale FO/RO system was built [53]. At full-scale operation (Fig. 7), landfill leachate was collected and pretreated using hydrochloric acid to prevent inorganic scaling. The system consisted of four treatment trains, each with six FO plate-and-frame membrane stacks in series. While the leachate became concentrated, diluted draw solution was treated and reconcentrated using an RO system, producing high-quality permeate meeting discharge regulations [53].

After approximately one year of operation, the full-scale system treated 18,500 m³ (~5 million gallons) of landfill leachate at greater than 91% water recovery. The FO/RO process also continually produced permeate having less than 100 mg/L TDS. Contaminants of concern, including cadmium, chromium, copper, lead, zinc, and ammonia were consistently more than 90% rejected, with effluent biochemical oxygen demand (BOD) concentrations below 5 mg/L. The FO/RO application successfully provided effluent contaminant concentrations lower than the National Pollutant Discharge Elimination System (NPDES) total maximum daily loads (TMDL) [23,53].

3.2.2. FO treatment of centrate from anaerobic digesters

Municipal wastewater treatment facilities typically treat primary and secondary biosolids in aerobic or more commonly anaerobic digesters. Solids digestion promotes degradation of organic constituents and BOD, producing stabilized biomass and biogas. After digestion the sludge is dewatered, producing a concentrated liquid waste stream (i.e., centrate) and dewatered biosolids. While the biosolids are typically land applied or trucked for off-site disposal, the liquid waste stream is commonly returned to the facility headworks. This practice increases facility loading because the liquid contains high nutrient concentrations (e.g., organic nitrogen, orthophosphate, ammonia), dissolved metals, TDS, total suspended solids (TSS), and organic carbon [23,54,55]. By removing this return stream, treatment facilities can reduce total waste loadings, operating costs, and effluent nitrogen and phosphorous concentrations. Concentrated centrate can also be sold as a product and used as a fertilizer.

An FO treatment study [54] was conducted at the Truckee Meadows Water Reclamation Facility in Reno, Nevada in 2006 as a method to treat and reduce the volume of centrate produced at the facility. The purpose of the study was to evaluate FO performance for concentrating raw and filtered centrate as an alternative to their common practices. During the bench-scale investigation centrate was filtered through a 150-mesh sieve prior to the FO process. Water was then extracted from the filtered centrate across a CTA FO membrane operating in osmotic dilution mode [27]. FO provided sustainable water flux and high rejection of contaminants of concern while successfully concentrating raw and filtered anaerobic digestor centrate. While water flux decline was noticed between each test cycle due to fouling, membrane cleaning restored water flux to its original level. Even though increased flux decline was observed when testing raw centrate, the ability to recover most permeate flux indicated that minimal irreversible fouling occurred during the FO process.

Constituents of concern included ammonia, total Kjeldahl nitrogen (TKN), and ortho-phosphate with average feed concentrations of 1300, 1400, and 240 mg/L, respectively. FO provided 87% ammonia rejection, 92% TKN rejection, and greater than 99% rejection of phosphorous, color, and odor compounds. Results from the study suggest that combining the FO process with RO could successfully produce 35,000 gal/day (130 m³/day) of purified water from a 50,000 gal/day (190 m³/day) stream of centrate [54].

3.2.3. FO treatment of domestic wastewater and osmotic membrane bioreactors (OMBR)

Stringent wastewater treatment regulations and advancements towards indirect and direct potable water reuse require implementation of improved treatment processes to produce high quality reclaimed water. Membrane bioreactors (MBR) have demonstrated the ability to provide advanced treatment, producing effluent suitable for irrigation, industrial processes, and even potable water when provided proper effluent polishing [56]. MBRs replace the combined biological, clarification, and filtration processes in conventional, municipal wastewater treatment facilities. Using MF or UF membranes, MBRs reject nearly all suspended solids and maintain high biomass concentration, providing consistent effluent quality in a significantly smaller footprint than traditional treatment processes (i.e., sequencing batch reactors, extended aeration facilities, lagoons) [56].

Yet, due to limited rejection of TDS, low molecular weight contaminants, and trace organic compounds (TOrCs), and because of membrane properties and fouling propensity associated with the operation of conventional MBRs, FO has been investigated as a potential alternative for advanced wastewater treatment [56–66]. Independent studies conducted since 2008 have aimed at developing an efficient osmotic membrane bioreactor (OMBR). For example, Achilli et al. [56] investigated membrane fouling, water flux, reverse solute diffusion, and nutrient rejection at the bench scale. Three flat-sheet CTA membranes were employed in a plate and frame cell configuration and results concluded that membrane fouling in the OMBRs was lower than in MF/UF MBRs. Water flux was restored to within 10% of the original flux using membrane relaxation (when no filtration takes place) and osmotic backwashing, showing minimal irreversible fouling. Flux was easily sustained throughout the duration of the experiments, and decline in the driving force was associated with easily cleaned fouling layers. The FO membranes rejected 99% of influent TOC and 98% of influent ammonia. This is significantly better rejection than that of porous MBR membranes,
which can range between 28 and 87% of soluble organic matter, depending on the extent of membrane fouling.

Another important study by Alturki et al. [57] was recently published, evaluating FO rejection of TOrCs that pass through MBR processes. A thorough literature review revealed that conventional wastewater treatment processes do not provide effective removal of TOrCs. MBRs provide slight enhancement of pollutant removal through biological degradation due to increased solids retention times and biomass concentration. However, due to the porous nature of MBR MF and UF membranes, low molecular weight TOrCs can easily flow through the treatment process. Only those pollutants readily biodegradable and hydrophobic are removed. Flat sheet cellulose acetate membranes were employed in a plate and frame test cell and 50 TOrCs, each with an average concentration of 750 ng/L, were investigated. Experimental results show that the OMBR provided high rejection (permeate concentration below analytical detection limits) of many TOrCs with molecular weights greater than 266 g/mol. This high rejection promoted biological degradation of the pollutants within the bioreactor. Rejection of pollutants smaller than 266 g/mol was highly variable, ranging from minimal rejection to removal below analytical detection limits.

Long-term pilot-scale tests have since been conducted using a novel FO plate and frame membrane module (Fig. 8) [67]. The objective of the long-term pilot scale evaluation was to determine the sustainability and permeate water quality of a coupled FO and RO process. High quality permeate was consistently produced for an additional four months of system operation and the salinity in the activated sludge was sustained at a low concentration. FO permeate water quality of a coupled FO and RO process. High quality permeate water quality was provided for 125 days of operation without membrane cleaning.

**3.2.4. FO for concentration of foods and beverages**

The concentration of liquid foods and beverages is an important and equally sensitive process in industrial food production. Traditionally, foodstuffs are concentrated using multi-stage vacuum evaporation or even RO. However, these processes can reduce the quality of the final concentrated product. Heat generation and vapor losses can negatively impact food color, taste, and potentially the nutritional value of the final concentrate [70], and RO operation is limited by osmotic pressures at high feed concentrations. Jiao et al. [71] and Petrotos and Lazarides [70] have published thorough summaries of membrane application in the food industry, including results from FO studies. The first attempt to use modern applications of FO was by Popper et al. in 1966 [72].

First generation RO membrane made of cellulose acetate was used in both flat sheet and tubular configurations. Using a highly concentrated NaCl draw solution, the membranes produced 2.5 L/m² h and were able to increase grape juice concentration by 44° Brix (the measure of sugar content of an aqueous solution). However, reverse solute diffusion of salt [73,74] into the grape juice concentrate demonstrated the need for different, more appropriate draw solutions. Improving upon the concept, Beaudry and Lampi [75] investigated a 72° Brix sugar draw solution employed in a newly developed plate-and-frame membrane element, housing a thin film composite (TFC) membrane coupon. These improvements increased water flux to 5–6 L/m² h, while providing greater than 99.9% rejection of orange juice acids and red raspberry juice color sensory. In 1993, Wrolstad et al. [76] compared Osmotek’s FO treatment of red raspberry juice to traditional vacuum concentration. Using a high fructose corn syrup draw solution, the resulting FO concentrate was analyzed and found to be of equal or higher quality than that produced by vacuum evaporation.

Two studies conducted by Petrotos et al. [72,77,78] further investigated these findings when applying FO to tomato juice concentrate. This is a very challenging application because tomato juice is considered one of the most concentrated vegetable juices in the industry. Experimental results suggest that draw solution viscosity directly impacts overall water flux (e.g., low viscosity draw solutions provide improved water flux). Additionally, it was concluded that decreasing membrane thickness provided an exponential increase in water flux. When the tomato juice feed stream was also pretreated with a filtration process FO performance improved, providing a 39% increase in water flux in comparison to no pretreatment. Over ten years later, FO is still being investigated for treatment of liquid foodstuff, where Garcia-Castelloa et al. [79] concentrated orange peel press liquor using FO. This research...
showed that FO is a promising alternative to traditional dewatering processes and also concluded that minimal pretreatment prior to FO may help limit declining permeate flux due to membrane fouling.

Based on tested membrane performance, FO can be a well-suited treatment alternative for use in the food processing industry and competitive with traditional vacuum evaporation and RO. Under optimized membrane design and proper choice of osmotic draw solution, sustainable water flux can be generated at low temperatures and low pressures that are desired in these types of applications.

4. Technological progress to enable better utilization in the O&G and other industries

FO treatment has shown great applicability and competitiveness in challenging industrial applications. Two commercialized FO membrane processes have proven successful in treatment of O&G wastewater for beneficial water reuse. Nonetheless, to better apply the treatment strategies and optimize system performance, substantial improvements can still be made in FO. Three independent reviews[23,25,48] presented several shortcomings of FO that need to be addressed by future research and development. Membrane manufacturing and module design are being continually improved, including increasing membrane robustness, permeability, chemical stability and range and rejection of contaminants of concern. New FO membranes for O&G must minimize internal concentration polarization [50,74,80–85] in order to reduce the loss of osmotic driving force across the membrane as waste streams become concentrated. Improvements should also be made to draw solutions, maximizing osmotic pressure while minimizing reverse solute diffusion, regeneration and recovery costs, toxicity, and reactivity with the membrane active layer.

4.1. New membranes

First generation FO membranes were produced by HTI using cellulose triacetate. This polymer is cast with an embedded polyester mesh for membrane support while forming a dense semi-permeable active layer. The goal was to minimize the active layer thickness of the asymmetric membrane, theoretically increasing membrane water permeability without compromising contaminant rejection or membrane integrity. These CTA membranes for FO are still under development and are the workhorse of the Green Machine; however, studies and recent field tests in regional gas plays have shown that while these first generation membranes are very robust, they do not have the desired water permeability and salt rejection, and they can only operate in a narrow pH range [86,87]. Recently developed TFC FO membranes by HTI and Oasys for this same application were tested by Coday et al. [88]. The TFC membranes exhibited better water flux than CTA membranes; however, the reverse salt flux of TFC membranes was higher and more affected by the transmembrane pressure common in the latest O&G FO treatment membrane modules. Rejection of organic molecules was comparable between the TFC and CTA membranes, at approximately 96%. The study demonstrated that new membrane materials and structure, coupled with operating conditions, might influence the preferential reverse diffusion and rejection of charged ions. This phenomenon is important and can impact specific process applications and requires further investigation.

Looking to the future, Wang et al. [86] suggest that the most effective FO membranes must have a very thin active layer supported by a thin support whose structure is highly porous to minimize internal concentration polarization. The membrane surface composition should be hydrophilic, which may help minimize O&G foulant deposition on the membrane surface and increase water permeability when treating viscous fracturing flowback fluids. Furthermore, the membrane chemistry must tolerate large shifts in pH and maintenance with various aggressive chemicals to maintain efficient and uninterrupted operation at the well site.

Several academic studies have focused on the advanced development of these FO membranes [82,83,85–87,89–96]. For example, Wang et al. [86] investigated the production of thin-film composite FO hollow fibers with an ultra-thin active layer. This active layer, very similar to an RO selective layer, can be cast on either the inside or outside of the hollow fiber membrane wall. Results from the experiments suggest that it is possible to easily tailor this process and the membrane active layer to meet specified requirements. The use of hollow fiber membranes could increase membrane-packing density and avoid the severe pressure drop of spiral wound membrane modules when they become fouled/clogged. Qiu et al. [87] produced a positively charged flat sheet membrane using polyamide-imide (PAI) substrate with a polyelectrolyte post-treatment. This produced an asymmetric, micro–porous membrane with an active layer similar to that of a NF membrane. Unfortunately, membranes for O&G wastewater treatment should be negatively charged, which would decrease the affinity of negatively charged organic molecules to adhere to the membrane surface. Setiawan et al. [89] built upon this same research to develop a PAI membrane with a less positively charged active layer to help mitigate the attraction of negatively charged organic molecules. In general, both casting techniques and membrane substrate selection have allowed polymer scientists to produce better FO membranes, tailored for specific
applications and different feed water compositions, with the goal of satisfying the criteria established above. For O&G wastewater recovery, membrane manufacturers are challenged with balancing several requirements; membrane robustness, support layer porosity, and rejection should be maximized, while the thickness of the membrane active layer and tortuosity of the membrane’s support layer should be minimized.

4.2. New membrane configurations

Several different membrane configurations have been developed and investigated in attempts to provide the best overall rejection, water flux, and operating efficiency, given certain feed water composition and characteristics. Well-developed configurations include plate-and-frame, spiral wound, hollow fiber, and tubular. FO membranes have been tested at the pilot-scale using traditional spiral wound and plate-and-frame configurations. Spiral wound modules are very similar to those used in traditional RO applications but with specific design modifications [23]. Typical RO spiral wound modules only accept one stream flow (e.g., feed stream) while FO modules must accept two streams simultaneously (Fig. 10). To do this, the membrane envelope and center collection tube in spiral wound FO modules are modified. The center tube (now the draw solution conduit) is plugged half way and the envelope is partially glued down the centerline [23]. This forces the draw solution to enter one half of the membrane envelope, flow across the membrane surface, and be collected in the other half of the plugged center tube. The feed solution flows through the module over the modified membrane envelopes, similar to the feed flow in RO spiral wound modules. Using spiral wound modules, feed channel clogging has been observed in previous O&G tests in the field; this is especially true when no pretreatment is applied before the FO process as practiced in the operation of the Green Machine.

In plate-and-frame modules, flat membrane sheets are held in place on frames and support systems. This system is typically submerged in a tank containing the feed stream (e.g., OMBR applications) while draw solution flows between the sealed membranes and plate support. This configuration can also be applied to O&G wastewater recovery; however, the footprint of the setup would likely increase in comparison to a spiral wound configuration. A custom tank may also be necessary that is capable of providing air scouring between the membrane plates, similar to an OMBR application.

Tubular and hollow fiber membranes (Fig. 11) are similar to MF and UF designs commonly employed in MBR applications. These configurations are durable and self-supported, with an active layer that can be produced on either the inner or outer sides of the tube/fiber. It is important to note that in hollow fiber membranes and other configurations, the orientation of the FO membrane (e.g., active layer facing feed or draw solution) can have significant impacts on system performance and fouling tendency.

For O&G wastewater treatment, one of the main technological challenges is the need to improve process hydraulics to avoid clogging of flow channels in the membrane elements and optimize the manufacturing and operation of membranes. Dissolved and suspended constituents in drilling and frac flowback wastewater and produced water are major membrane foulants, and upon concentration during the treatment process they can clog the membrane elements. Membrane fouling results in high operating and maintenance costs, prolonged system shutdowns, and ultimately permanent membrane damage. Although typical FO membranes are hydrophilic, and thus reduce fouling propensity of the membrane, precipitation of solids in the feed flow channels inside the membrane elements may retard the performance of the process. Novel membrane feed spacers, new membrane configurations (such as capillary membranes), optimized membrane manufacturing and incorporation of applicable pretreatment processes should be further investigated.

4.3. New draw solutions

Another important aspect to successful FO is the selection of a suitable draw solution that is well matched to the process (e.g., toxic or saline solutions are inadequate for beverage concentration). NaCl is used in the Green Machine because it is readily available on site, highly soluble, non-toxic, inexpensive, and easily reconstituted while providing high osmotic pressures. However, there are more than 500 inorganic compounds that can be potentially used as draw solutions; 14 were chosen and investigated in a recent study by Achilli et al. [42]. Other investigations have studied the applicability of dissolved gasses such as sulfur dioxide and ammonium bicarbonate, or even nanoparticles such as magnetoferritin as suitable draw solutions in tailored FO applications [18,42–47]. A summary of FO draw solutions is provided in Table 2. Due to the highly saline nature of O&G produced water, which effectively lowers the osmotic pressure driving force, innovative draw solutions with exceptionally high osmotic pressure are required. Solutions that are also compatible with O&G reuse options such as hydraulic fracturing or well drilling must also be considered.
4.4. Pretreatment before FO

Similar to pressure driven membrane processes, FO can be appreciably enhanced if appropriate pretreatment processes are implemented upstream of the FO process. While the Green Machine operates with no pretreatment, the MBC relies on adequate feed stream pretreatment to protect the FO membranes. The following is a list of suitable pretreatment processes, each with a short description:

- Coagulation/flocculation aids in the removal of suspended and colloidal particulates to reduce premature membrane fouling. It can make the fouling cake layer more porous and permeable when treating highly fouling feed streams, such as O&G frac flowback and concentrated domestic wastewater sludge and digester centrate.
- Acid/base pH control aids in precipitation of dissolved metals and precipitated particulates to reduce premature membrane fouling. It can make the fouling cake layer more porous and permeable when treating highly fouling feed streams, such as O&G frac flowback and concentrated domestic wastewater sludge and digester centrate.
- Scale mitigation/softening aids in precipitation or exchange of scale forming compounds to limit premature membrane scaling. Scale mitigation can be used prior to FO when treating high salinity produced waters (>70,000 ppm) having elevated concentrations of sparingly soluble salts.
- Filtration (granular media/MF/UF) aids in the removal of suspended and precipitated particulates to reduce premature membrane fouling and flow channel blocking/clogging. Filtration can be used when treating grit laden drilling mud and frac flowback wastewaters. Filtration may be especially used to protect TFC FO membranes because the active layer of these membranes is more delicate than that of the CTA FO membrane.
- Dissolved air flotation aids in the removal of oil, fats, and insoluble organics to reduce premature membrane fouling and damage to the membrane active layer. Dissolved air flotation can be used to remove elevated concentrations of emulsified hydrocarbons from produced waters, which may sorb to the membrane active layer and irreversibly foul the membrane.
- Advanced oxidation aids in the destruction of oils and fats and oxidizes reduced inorganic metal species for subsequent removal. Advanced oxidation can be especially important when treating frac flowback wastewater by further degrading any remaining polymers or guarins remaining from the hydraulic fracturing process.
- Disinfection minimizes the potential for biological fouling and degradation of the membrane active layer.

5. Conclusions

Forward osmosis is a suitable and effective process for treatment of difficult waste streams, and it was demonstrated at all scales of research and development, including bench scale, pilot scale, and commercial demonstration. Specifically in O&G exploration and production, FO is a promising technology that enables exploration companies to utilize effective wastewater treatment and promote beneficial water reuse in decentralized and remote locations. Currently, there are several different approaches and methods for implementing the technology; however, it is unclear which approach is most suitable, leaving significant room for more research. Ultimately, O&G exploration companies will choose the water management option that (a) is physically practical for on-site operation with their waste stream, (b) accepted by state and federal regulators, and (c) sustainable over extended periods of operation.

Compared to traditional disposal methods, both the HTI Green Machine and the Oasis Water Membrane Brine Concentrator FO systems demonstrated net cost advantages of more than 45% and 60%, respectively, in recent demonstration scale tests; however, a direct cost comparison between these two FO technologies is difficult to conduct at this time. The Green Machine is suitable for treating O&G waste streams with minimal or no pretreatment, but is currently more suitable for treating wastewaters with less than 70,000 ppm TDS. The Membrane Brine Concentrator system uses two pretreatment steps prior to FO membrane treatment, but can target feed stream salinities in excess of 70,000 ppm TDS.

FO has shown great versatility by successfully treating a wide range of feed stream salinities and producing equally wide ranges of product water quality—from diluted saline solution to RO permeate suitable for potable and non-potable reuse. Ultimately, other industries that produce complex liquid streams can benefit from the experiences of FO treatment of O&G E&P wastewater. The limitations of FO need further investigation, as new generations of TFC membranes and novel draw solutions are being developed. Further research is needed to test these membranes and draw solutions in conjunction with true wastewater streams to determine if they can further enhance the FO process for these difficult applications.
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